

**Research article** 

**Open Access** 

## Analytical method development and validation for the simultaneous estimation of atomoxetine hydrochloride by using rp-hplc technique

CH. Sai Teja, M. Sandeep, L. Shekar, Sheshagiri Palthi, P. Shireesha, Beaula Rani\*

Department of pharmaceutical Analysis, Teegala Ram Reddy College of Pharmacy, Telangana, India

#### **Corresponding author: Beaula Rani**

# ABSTRACT

The proposed methods were validated. The accuracy of the methods was assessed by recovery studies at three different levels. Recovery experiments indicated the absence of interference from commonly encountered pharmaceutical additives. The method was found to be precise as indicated by the repeatability analysis, showing %RSD less than 2. All statistical data proves validity of the methods and can be used for routine analysis of pharmaceutical dosage form. It can be effectively applied for routine analysis in research institutions, quality control department in meant in industries, approved testing laboratories studies in near future.

Keywords: Atomoxetine, repeatability analysis, high resolution, retention time

## **INTRODUCTION**

Pharmaceutical analysis simply means analysis of pharmaceuticals. Webster' dictionary defines а pharmaceutical is a medical drug. A more appropriate term for a pharmaceutical is active pharmaceutical ingredient (API) or active ingredient to distinguish it from a formulated product or drug product is prepared by formulating a drug substance with inert ingredient (excipient) to prepare a drug product that is suitable for administration to patients. Research and development (R&D) play a very comprehensive role in new drug development and follow up activities to ensure that a new drug product meets the established standards is stable.

### **REVIEW OF LITERATURE**

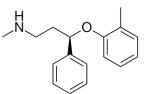
Gurmeet Chhabra, Chandraprakash Jain, Saurabh K Banerjee: A simple, reliable, specific, accurate and precise reverse phase high performance liquid chromatographic method was developed and validated for the determination of Atomoxetine HCl in pharmaceutical dosage form. Chromatographic separation was carried on RP-C8 column (Phenomenex, size:  $250 \times 4.60$  mm, particle size  $5\mu$ m) with a **Structure:**  mobile phase composed of acetonitrile and 10 mM disodium hydrogen phosphate buffer with 0.1% TEA (pH 3.0, adjusted with OPA) (55:45, v/v) in isocratic mode at a flow rate of 1mL/min. The detection was monitored at 271nm.

Zubaidur Rahman, Vijey Aanandhi M, Sumithra M: A simple, novel, sensitive, rapid high-performance liquid chromatographic (RP-HPLC) method has been developed and validated for quantitative determination of atomoxetine HCl (ATH) in bulk and formulations. The chromatographic development was carried out on RP-HPLC. The column used as Xterra RP 18 (250 mm — 4.6 mm, particle size), with mobile phase consisting of methanol: water 80:20 V/V. The flow rate was 1.0 mL/min and the effluents were monitored at 270 nm. The retention time was found to be 5.350 min.

#### DRUG PROFILE

#### Atomoxetine

Atomoxetine is a nonstimulant medication marketed in the form of the R (-) isomer as this structure seems to have approximately nine-fold more potency than the S (+) isomer. It is a phenylpropanolamine derivative that presents a similar structure to the tricyclic antidepressants.



**IUPAC Name**: (3*R*)-*N*-Methyl-3-(2-methylphenoxy)-3-phenylpropan-1-amine **Molecular Formula**: C<sub>17</sub>H<sub>21</sub>NO **Molecular Weight**: 255.3547

## AIM

To develop and validate new HPLC method for Atomoxetine in pharmaceutical dosage form.

## **MATERIALS AND METHODS**

Instruments used					
UV-Visible Spectrophotometer	Nicolet evolution 100				
UV-Visible Spectrophotometer software	Vision Pro				
HPLC software	Lab Solution				
HPLC	SHIMADZU 2010				
Ultra sonicator	Citizen, Digital Ultrasonic Cleaner				
pH meter	Global digital				
Electronic balance	Mettler Toledo				
Syringe	Hamilton				
HPLC Column	Inertsil ODS 3V(150x4.6mm) 4µm				

#### **Reagents used**

Water	HPLC Grade			
Methanol	HPLC Grade			
Potassium Dihydrogen Phosphate	AR Grade			
Acetonitrile	HPLC Grade			
Dipotassium hydrogen phosphate	AR Grade			
Orthophosphoric acid	HPLC Grade			

Drugs used					
Atomoxetine (API)	Gift Samples obtained from Chandra labs, Hyd.				
Strattera (60 mg)	Obtained from local pharmacy				

#### Mobile Phase Preparation Sodium Buffer

About 0.5g of 1-Heptane Sulphonic acid and 0.5g of Sodium chloride was weighed and transferred into 1000ml volumetric flask and add sufficient water for dissolving and make upto mark with water, sonicate for 15mins. Add 2.5ml of triethylamine and adjust the  $p^{\rm H}$  to 3.8 using orthophosphoric acid and filter the solution through 0.45micron membrane filter.

#### Mixed Phosphate Buffer

Weigh 2.95g of Potassium Dihydrogen Ortho Phosphate and 0.5g of Dipotassium Hydrogen Ortho Phosphate into 1000ml volumetric flask and add sufficient water for dissolving, then make upto mark with water, sonicate for 15mins. Add 2.5ml of triethylamine and adjust the  $p^{H}$  to 3.5 using orthophosphoric acid and filter the solution through 0.45micron membrane filter.

### METHOD DEVELOPMENT AND VALIDATION Introduction to Method Development

The number of drugs introduced into the market is increasing every year. These drugs may be either new entities or partial structural modification of the existing one. Often a time lag exists from the date of introduction of a drug into the market to the date of its inclusion in pharmacopoeias.

## Method Development Using HPLC

In method development, an attempt to select the best chromatographic conditions like the best column, the best mobile phase, the detection wavelength etc. to be used for routine analysis of any drug is done. For the method development by HPLC method some information about the sample is very essential.

Beaula Rani et al / J. of Pharmacreations Vol-9(3) 2022 [186-190]

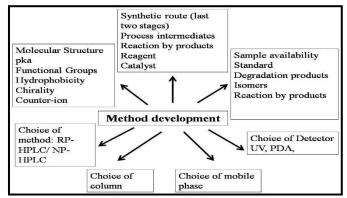


Fig 1: Outline of the process involved in method development

# Method Validation (ICH Guidelines)

## Accuracy

The accuracy of an analytical procedure expresses the closeness of agreement between the value which is accepted either as a conventional true value or an accepted reference value and the value found. Accuracy should be established across the specified range of the analytical procedure.

#### Precision

The precision of an analytical procedure expresses the closeness of agreement (degree of scatter) between a series of measurements obtained from multiple sampling of the same homogeneous sample under the prescribed conditions. Precision may be considered at three levels: repeatability, intermediate precision and reproducibility.

#### Specificity

Specificity is the ability to assess accurately the analyte in the presence of components which may be expected to be present in the sample matrix. Typically, these might include impurities, degradants, matrix, etc. it is a measure of the degree of interference from such other things such as other active ingredients, excipients, impurities, and degradation products, ensuring that a peak response is due to a single component only.

## Limit of Detection (LOD)

The detection limit of an individual analytical procedure is the lowest amount of analyte in a sample which can be detected but not necessarily quantitated as an exact value. It is a limit test that specifies whether or not an analyte is above or below a certain value.

## Limit of Quantitation (LOQ)

The limit of quantitation (LOQ) is defined as the lowest concentration of an analyte in a sample that can be determined with acceptable precision and accuracy under the stated operational conditions of the method.

## Linearity and Range

Linearity is the ability of the method to elicit test results that are directly proportional to analyte concentration within a given range. Linearity is generally reported as the variance of the slope of the regression line. Range is the (inclusive) interval between the upper and lower levels of analyte that have been demonstrated to be determined with precision, accuracy, and linearity using the method.

#### Robustness

Robustness is the capacity of a method to remain unaffected by small deliberate variations in method parameters. The robustness of a method is evaluated varying method parameters such as percent organic solvent, pH, ionic strength, or temperature and determining the effect (if any) on the results of the method.

#### System Suitability

System suitability tests are an integral part of gas and liquid chromatographic methods. They are used to verify that the resolution and reproducibility of the chromatographic system are adequate for the analysis to be done. The tests are based on the concept that the equipment, electronics, analytical operations and samples to be analyzed constitute an integral system that can be evaluated as a whole.

## **RESULTS AND DISCUSSION**

## Solubility Studies

These studies are carried out at 25 °C

#### **Atomoxetine**

Freely soluble in Methanol, Acetone and insoluble in Water

#### Determination of Working Wavelength ( $\lambda$ max) Preparation of standard stock solution of Atomoxetine

10 mg of Atomoxetine was weighed and transferred in to 100ml volumetric flask and dissolved in Methanol and then make up to the mark with methanol and prepare 10  $\mu$ g /ml of solution by diluting 1ml to 10ml with Methanol.

#### **Results**

The wavelength of maximum absorption ( $\lambda_{max}$ ) of the drug, 10 µg/ml solution of the drug in Methanol were scanned using UV-Visible spectrophotometer within the wavelength region of 200–400 nm against Methanol as blank. The resulting spectra are shown in the 8.3 and the absorption curve shows characteristic absorption maxima at 253 nm for Atomoxetine, selected as detector wavelength for the HPLC chromatographic method.

#### Beaula Rani et al / J. of Pharmacreations Vol-9(3) 2022 [186-190]

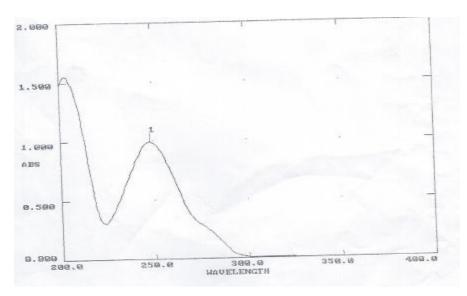


Fig 2: UV-VIS spectrum of Atomoxetine at 253nm

## **METHOD DEVELOPMENT OF ATOMOXETINE**

#### Preparation of standard solution

weigh accurately 10 mg of Atomoxetine in 25 ml of volumetric flask and dissolve in 25ml of mobile phase and make up the volume with mobile phase. From above stock solution 20  $\mu$ g/ml of Atomoxetine is prepared by diluting 1.5ml to 10ml with mobile phase. This solution is used for recording chromatogram.

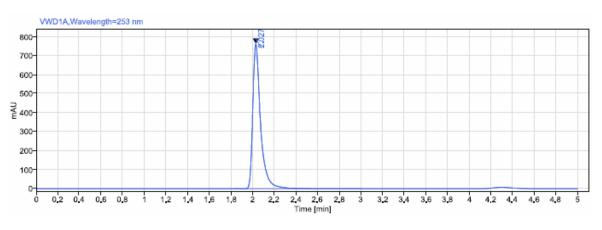


Fig 3: Chromatogram of Atomoxetine

#### Assay

#### **Preparation of samples for Assay**

**Preparation of standard solution:** weigh accurately 10 mg of Atomoxetine in 25 ml of volumetric flask and dissolve in 25ml of mobile phase and make up the volume with mobile phase. From above stock solution 20  $\mu$ g/ml of Atomoxetine is prepared by diluting 0.5ml to 10ml with mobile phase. This solution is used for recording chromatogram.

**Sample preparation:** weigh accurately 10 Tablets (Atomoxetine -60 mg) weigh accurately 10 mg of Atomoxetine in 25 ml of volumetric flask and dissolve in 25ml of mobile phase and make up the volume with mobile phase. From above stock solution 20  $\mu$ g/ml of Atomoxetine is prepared by diluting 0.5ml to 10ml with mobile phase. This solution is used for recording chromatogram.

#### VALIDATION HPLC METHOD VALIDATION System Suitability& System precision

To verify that the analytical system is working properly and can give accurate and precise results were evaluated by  $20\mu g/mL$  of ATOMOXETINE was injected six times and the chromatograms were recorded for the same.

#### **Method precision**

Method precision was determined by injecting sample solutions of concentration ATOMOXETINE ( $20\mu g/mL$ ) for six timesare prepared separately. The %RSD of determinations of ATOMOXETINE found to be within the acceptance criteria (less than 2.0%). %Assay Also within the limits (95.0 to 105.0) hence method is precise.

## Linearity and range

#### Preparation of standard stock solution

Standard stock solutions of ATOMOXETINE were prepared by dissolving 100 mg of ATOMOXETINE in 100 mL of Diluent. After that filtered the solution using 0.45-micron syringe filter and Sonicated for 5 min.

The relationship between the concentration (in %) of ATOMOXETINE and area of ATOMOXETINE should be linear in the specified range and the correlation should not be less than 0.99. The correlation coefficient for linear curve obtained between concentration vs. Area for standard preparation 0.990.

#### **Specificity**

A study to establish & determine the interference of blank and placebo as conducted. Analysis was performed on placebo in triplicate equivalent to about the weight of placebo in portion of test preparation as per test method. Chromatograms of blank and placebo solutions had shown no peaks at the retention times of ATOMOXETINE

#### Accuracy

Accuracy of the method was determined by Recovery studies. To the formulation (reanalyzed sample), the reference standards of the drugs were added at the level of 50%, 100%, 150%. The recovery studies were carried out three times and the percentage recovery and percentage mean recovery were calculated for drug is shown in Table 7.28 & 7.29.

#### **Robustness**

The Robustness of the method was determined. The results obtained by deliberate variation in method parameters.

Chromatographic changes		Rt(min)	<b>Tailing Factor</b>	<b>Theoretical Plates</b>	%RSD for Standard
Flow rate (mL/min)	0.8	2.429	1.74	5550	0.028
	1.2	1.738	1.28	2042	0.036
Temperature (°C)	35	2.031	1.53	3852	0.04
- · ·	45	2.031	1.59	4436	0.33

#### **Table 9.7: Results for Robustness of ATOMOXETINE**

The tailing factor was found to be within the limits on small variation of flow rate and wavelength.

#### Intermediate Precision (Ruggedness)

Intermediate precision (also called within-laboratory or within-device in different days, different analysts) is a measure of precision under a defined set of conditions: same measurement procedure, same measuring system, same location, and replicate measurements on the same or similar objects over an extended period.

## DISCUSSION

A simple and selective HPLC method is described for the determination of Atomoxetine Chromatographic separation was achieved on a C18 column using mobile phase consisting of a mixture of 40 volumes of Methanol, 40 volumes of Acetonitrile and 20 volumes of Water with detection of 253 nm. Linearity was observed in the range 50-150  $\mu$ g /ml for

#### **REFERENCES**

Atomoxetine ( $r^2 = 0.990$ ) for the amount of drugs estimated by the proposed methods was in good agreement with the label claim.

## CONCLUSION

From the above experimental results and parameters it was concluded that, this newly developed method for the simultaneous estimation of Atomoxetine was found to be simple, precise, accurate and high resolution and shorter retention time makes this method more acceptable and cost effective and it can be effectively applied for routine analysis in research institutions, quality control department in meant in industries, approved testing laboratories studies in near future.

- 1. Chatwal RG, Anand KS. High-performance liquid chromatography. Instrumental methods of chemical analysis. 5th ed. Mumbai: Himalaya publishers, 2010; 2.570-629.
- 2. Sharma BK. High-performance liquid chromatography. Instrumental methods of chemical analysis. 24th ed; Goelpublishers: Meerut, 2005. p. 295-300.
- 3. Dong WM. HPLC instrumentation and trends. Modern HPLC for practicing scientists. USA, 2006; 5-10, 78-110.
- 4. Typical diagram of HPLC. Available from: http://www.comsol.com/stories/waters\_corp\_hplc\_systems/full/Hplc diagram.
- 5. HPLC solvent properties. Available from: http://www.sanderkok.com/techniques/hplc/eluotropic series extended.htm.
- 6. Swartz ME, Ira Krull S. Analytical method development. Analytical method development and validation. 1st ed. New York: Marcel Dekker, Inc, 2009; 17-80.
- 7. Satinder A, Dong MW. Method development and validation. Pharmaceutical analysis by HPLC. 15th ed. New York, 2005; 16-70.
- 8. Snyder RL, Kirkland JJ, Glajch LJ. Getting started. Practical method development. 2nd ed. New York, 1997; 30-100.
- 9. Selection of buffers. Available from: http://www.sigmaaldrich.com/etc/medialib/docs/Aldrich/General\_ Information/labbasics\_pg144.Par.0001.File.tmp/labbasics\_pg144.pdf.
- 10. ICH. Text on validation of analytical procedures, International Conference on Harmonisation, IFPMA, Geneva, 1995. Vol. A–1 to A–3. ICH Q2A. p. 2-3.